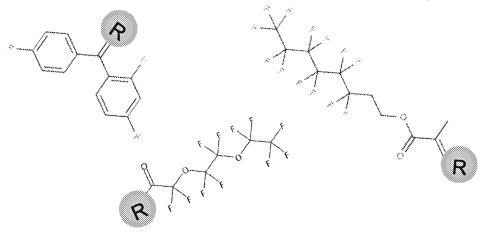
Per and Polyfluorinated Compounds: Health and Environmental Impacts Mark Strynar

Mark Strynar

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Highly Fluorinated Chemicals: A Sticky Issue

Webinar

April 19, 2017



Presentation Outline

- Introduce per- and polyfluoroalkyl substances (PFAS)
 - what are they?
- Discuss chemical properties of PFAS
 - why are they useful?
- Review what is known about routes of human exposure
- Some recent findings
- Describe animal and human health effects of PFAS
- Long-term outlook regarding PFAS



Major Considerations

The Commission of the European Communities: (2000) uses the Principle of Precaution ... "where there are threats of serious or irreversible damage, lack of full scientific certainty shall not be used as a reason for postponing cost effective measures to prevent environmental degradation".

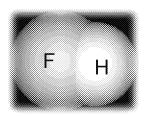
http://ec.europa.eu/dgs/health_consumer/library/pub/pub07_en.pdf

Green Chemistry...always strive to use the least toxic alternative available, with a preference for compounds that quickly and harmlessly degrade to their original starting materials.

Anastas PT, Warner JC. 1998. Green Chemistry: Theory and Practice. New York, NY Oxford University Press.

Per- and Polyfluoroalkyl Substances (PFAS)

- Synthetic analogs to long chain fatty acids but fluorine is used in the place of hydrogen
- PFAS are entirely manmade no natural sources and literally thousands of different formulations in use
- Many PFAS are extraordinarily persistent in the environment, cannot be broken down by natural systems





- PFAS are detected in all environmental media air, water, soil, sludge
- Like other persistent organic pollutants, many PFAS bioaccumulate in animals at the top of the food chain – birds, fish, livestock, and humans
- Environmental persistence leads to global distribution via air and water movement – releases here can be significant for communities on the other side of the world

Some Per- and Polyfluoroalkyl Substances (PFAS) Perfluorosuifonic acids Perfluorocarboxylic acids Fluorotelomer aicohol (ex. 8:2 FTOH) (ex. PFOA) (ex. PFOS) Perfluorophosphonic/phosphinic acids Perfluorosulfonamide (ex. If R=OH then PFOPA Perfluorinated cyclo sulfonates (ex. FOSA) If R=C8 perfluoroalkane then 8:8 PFPI) (ex. PFECHS) Perfluorosulfonamidoethanol Polyfluorinated ether carboxylates Fluoroteiomer phosphate esters (ex. N-EtPOSE) (ex. 4,8-dioxa-3H-perfluorononanoate) (ex. If R= OH then 8:2 monoPAP if R= 8:2 FTO ester then 8:2 diPAP) Polyfluorinated ether sulfonates Polyfluorinated polymeric unit (ex. Perfluoro (hexyl ethyl ether sulfonate)) (ex. 1H, 1H, 2H, 2H-perfluorodecyl acrylate)

Figure 1. Generic structures for polyfluorinated compounds. The n=8 linear carbon structures are shown for many of these examples, but n=4-14 linear and/or branched carbon units are generally possible.

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PFAS Are Used In Many Products



F F F F F F F OH

Textile treatments Paper coatings

Paints Surfactants

Pesticides Fire-fighting foam

Floor/Ski polish Photographic film

Denture cleaner Mining fluids

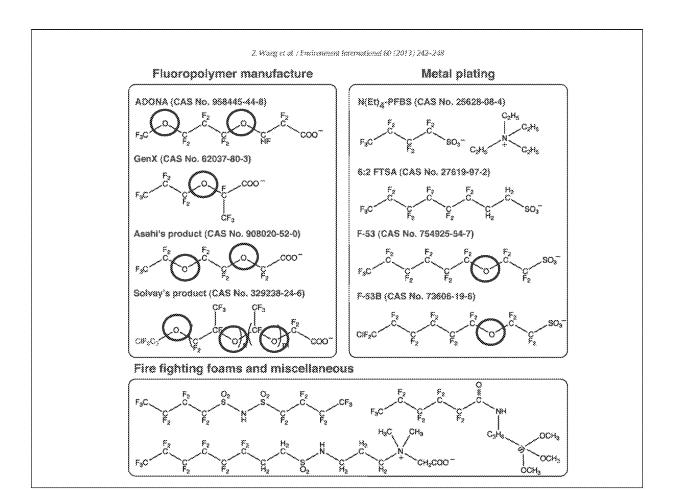
Polymers Non-stick cookware

Adhesives Caulks

Lubricant additive Carpets



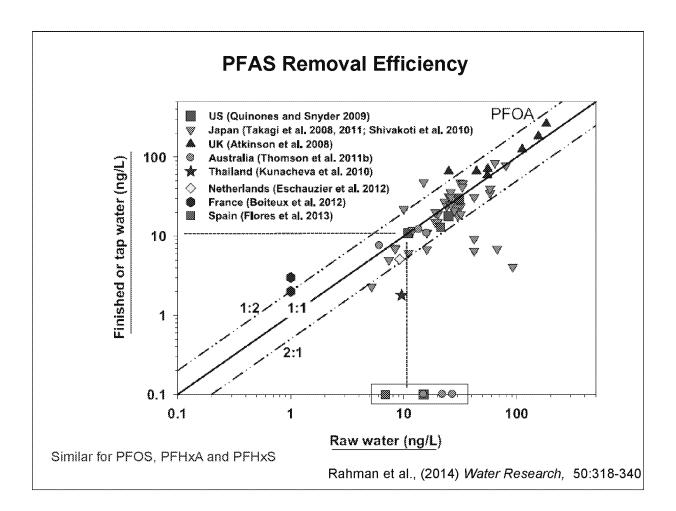






Sources of PFAS Exposure for Humans

- Best documented source is contaminated drinking water near industrial production facilities or waste disposal e.g., Cottage Grove, Minnesota; Parkersburg, West Virginia; Dalton, Georgia; Decatur, Alabama; Arnsberg, Germany; Osaka, Japan Lindstrom et al. 2011, Environ. Sci. & Technol. (45) 8015 8021
- * Food is also implicated in many studies, especially fish from contaminated waters, items contaminated by food packaging and breast milk Fromme et al. 2009, Inter. J. Hyg. & Envr. Heath (212) 239-270; Mogensen et al. 2015, Environ. Sci. & Technol. (49) 10466 10473
- House dust may also be an important route of exposure especially for children who ingest relatively higher levels of dust via hand-to-mouth activity Shoeib et al. 2011, Environ. Sci. & Technol. (45) 7999 8005
- Workplace exposures significant for some sectors: manufacturing or services making or directly using PFAS, apparel sales, waste treatment Nilsson et al. 2013 Environ. Sci.: Processes Impacts, 15, 814-822



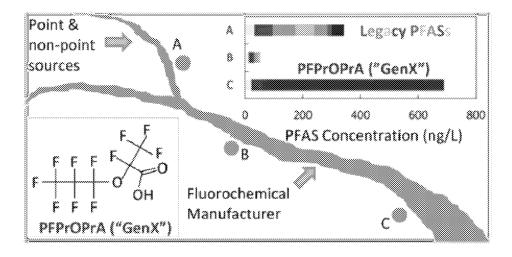
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Legacy and Emerging Perfluoroalkyl Substances Are Important Drinking Water Contaminants in the Cape Fear River Watershed of North Carolina

Mei Sun, $s^{-\frac{1}{2}}$ Elisa Arevalo, $s^{\frac{1}{2}}$ Mark Strynar, $s^{\frac{1}{2}}$ Andrew Lindstrom, $s^{\frac{1}{2}}$ Michael Richardson, $s^{\frac{1}{2}}$ Ben Kearns, Adam Pickett, $s^{\frac{1}{2}}$ Chris Smith, $s^{\frac{1}{2}}$ and Detlef R. U. Knappe $s^{\frac{1}{2}}$

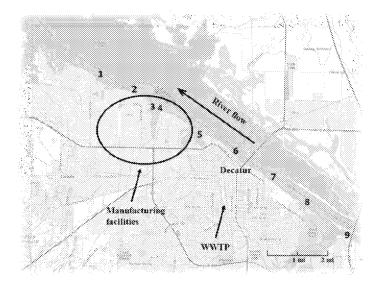




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Novel Polyfluorinated Compounds Identified Using High Resolution Mass Spectrometry Downstream of Manufacturing Facilities near Decatur, Alabama

Seth Newton, $^{\dagger \otimes}$ Rebecca McMahen, $^{\sharp}$ James A. Stoeckel, $^{\$}$ Michael Chislock, $^{\$}$ Andrew Lindstrom, † and Mark Strynar $^{\$,\dagger}$ \otimes





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Figure 3. Symbesis reaction for the production of a traditional perfluorinated carboxylic acid using tetrafluoroethene (A) and the hypothesized synthesis of the proposed polyfluorinated carboxylic acids using 1.1 difluoroethere. (B) where n = 2-10.

WWIP List 2ml



Fluorinated Compounds in U.S. Fast Food Packaging

Laurel A. Schaider,* * Simona A. Balan, * Arlene Blum, * David Q. Andrews, * Mark J. Strynar, * Margaret E. Dickinson, * David M. Lunderberg, * Johnsie R. Lang, * and Graham F. Peaslee*

Percent with fluorine Dessert & bread wrappers 56% Sondwich & burger 38% Propertioned 20% Properti

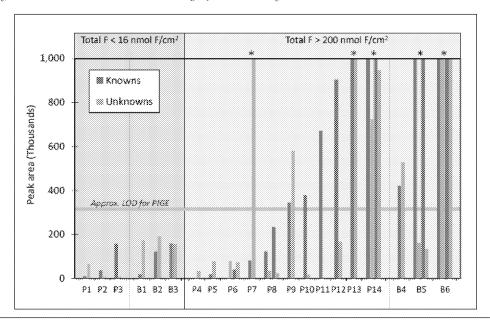
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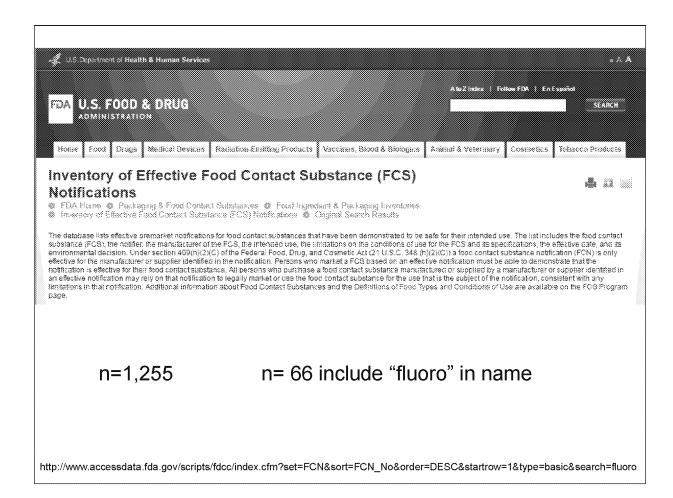


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"Fluoro" Substructures in Database

PFAS Health Effects Summary

 These chemicals not only persist in the environment, but also inside the body once ingested, particularly in humans. GenX data from Gannon et al., 2016

| Serum half-life | PFBS (C4) | PFHxS (C6) | PFOS (C8) | PFBA (C4) | PFHxA (C6) | GenX | PFOA (C8) | PFNA (C9) |
|--------------------|--------------|---------------|--------------|--------------|---------------|--------------------------|--------------|--------------|
| Mouse | 5 hr | 30 days | 40 days | 12 hr | 2 hr | alpha 5.2hr beta 31hr | 20 days | 60 days |
| Humans | 28 days | 8.5 years | 4-5 years | 3 days | 32 days | ??? | 3-4 years | unknown |

- Some of these chemicals are more potent than the others, but all of them have the similar effects (PPAR α activation)

| | PFBA (C4) | PFPeA (C5) | PFHxA (C6) | PFHpA (C7) | PFOA (C8) | PFNA (C9) | PFDA (C10) |
|--------|--------------|---------------|---------------|---------------|--------------|--------------|---------------|
| Mouse | 1.0 | 1.1 | 1.3 | 4.6 | 8.5 | 10.2 | 2.6 |
| Humans | 1.0 | 1.4 | 1.6 | 5.0 | 6.5 | 6.8 | |

Laboratory results suggest that PFAS effects are additive

Slide from Chris Lau, US EPA

Profiles of PFAS Toxicity and Adverse Health Effects

- Liver and Metabolic toxicity
 - Mouse: enlarged and fatty liver, decreased serum cholesterol, triglycerides
 - Humans: increased serum cholesterol, uric acid
- Reproductive and Developmental Toxicity
 - Mouse: neonatal mortality, low birth weight, growth deficits, developmental delays
 - Humans: preeclampsia, low birth weight and small size, delayed onset of puberty
- Tumor Induction
 - Mouse: liver, pancreas and testes
 - Humans: kidney and testes
- Immunotoxicity
 - Mouse: atrophy of thymus and spleen, suppressed immune responses
 - Humans: reduced immune responses to vaccines in children
- Endocrine Disruption
 - Mouse: reduced serum thyroid hormones
 - Humans: slight elevation of serum thyroid hormones
- Neurotoxicity
 - Mouse: a few reports of neuronal deficits and behavioral abnormalities
 - Humans: some reports of learning disability

Slide from Chris Lau, US EPA

Newer PFAS: Summary of what is Known?

Most compounds have limited but growing toxicology testing for some PFAS Gordon et al., (ADONA), Rae et al., (GenX), Dewitt et al., (GenX), Serex et al., 2014 (6:2 FTOH), Danish EPA 2015 (short-chain PFAS) BUT no information on others (Strynar et a., 2015, Newton et al., 2016, Schaider et al., 2017)



Toxicology found to be "more favorable" than longer chain PFAS (Dupont GenX brochure) but NOT non-toxic (Rae et al., 2015 (GenX); Gordon et al., (ADONA), Serex et al., 2014 (6:2 FTOH)

Shorter chain length PFAS are: almost completely absorbed orally (Chang et al., 2008 (PFBA), Olsen et al., 2009 (PFBS) and Gannon et al., 2011 (PFHxA) more rapidly eliminated in mammals Gannon et al., 2016 (GenX) and poorly attenuated in Traditional drinking water systems (Sun et al., 2016)

Environmental bio-concentration of shorter chain length is of low concern (Hoke et al., (GenX), BUT has been shown to occur in fish species (Chu et al., 2016., (FBSA)

Terminal per- and polyfluorinated metabolites are recalcitrant Danish EPA 2015 (short-chain PFAS), Wang et al., 2013 (PFPEs), ECHA 2015

Future Perspectives

- Demand for PFAS performance chemicals increasing with a shift in production of "legacy" materials to the developing world (India, Poland, China, Russia)
- New generation of "replacement" PFAS now being produced in the industrialized world, but their identity and health effects are relatively unknown.
- Environmental and health effects research on "replacement" PFAS now underway – preliminary results suggest they have similarity to legacy compounds
- New research on human exposure of PFAS and their adverse health effects, as well as their ecological impacts will support risk assessment and regulatory decisions
- Virtually every person has PFAS in their blood biomonitoring studies will inform the trends of change in the future regarding new and legacy chemicals



